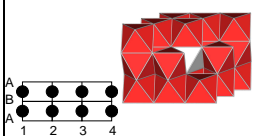
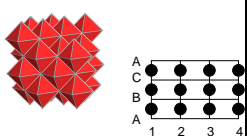
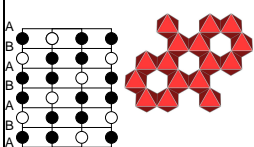
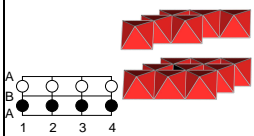
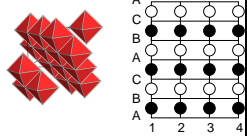
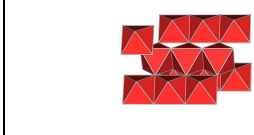
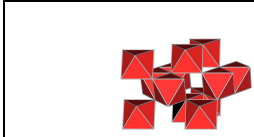

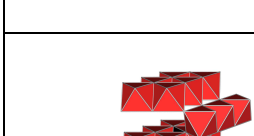
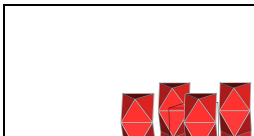


4.3. Dichte Packungen mit gefüllten Oktaederlücken

X hexagonal dicht	Formel	CN (X)	Lückenbesetzung	X kubisch dicht
 NiAs	MX	6	1:	 NaCl
 Korund	M ₂ X ₃	4	2/3:	-
 CdI ₂	MX ₂	3	1/2	 CdCl ₂
 CaCl ₂ (Rutil)	Raum- netz		1/2:	
 α-PbO ₂			1/2:	 Anatas (TiO ₂)
 α-AlOOH			1/2:	
NFe ₂			2/3 u. 1/3	
 ZrI ₃	MX ₃	2	1/3	
BiI ₃	Kette		1/3:	
RhF ₃	Schicht		2/3 u. 0	YCl ₃
	Raumnetz			
α-NbI ₄ (Kette)	MX ₄	2/1	1/4	NbF ₄ (Schicht)
α-Nb ₂ Cl ₁₀ (mol.)	MX ₅	2/1	1/5	U ₂ Cl ₁₀ (mol.)
				UF ₅ (Kette)
α-WCl ₆	MX ₆	1	1/6	-